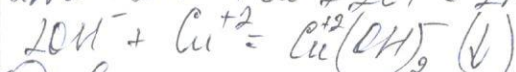
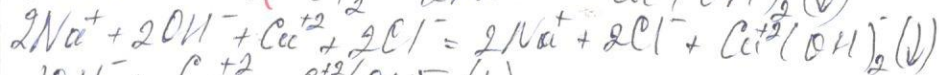
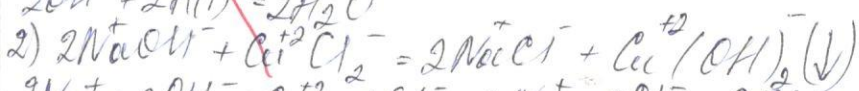
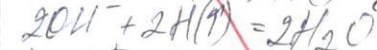
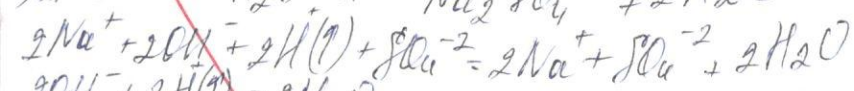
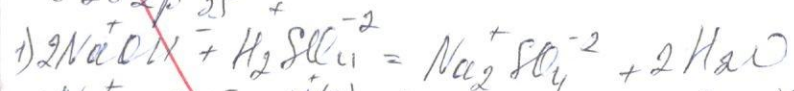


Винниченко Винниченко 9.5

①  $\text{Na}^+$   $\text{Mr} = 23$  и протон, 10 электронов, 7 нейтронов

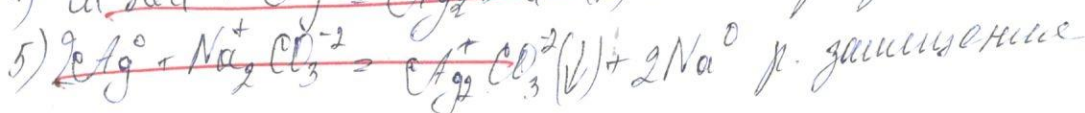
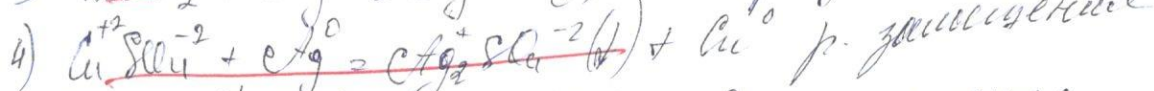
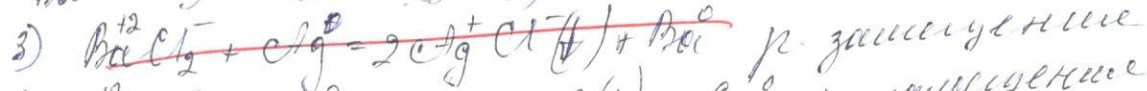
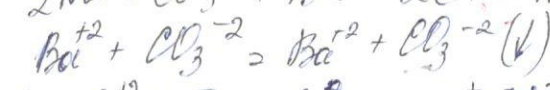
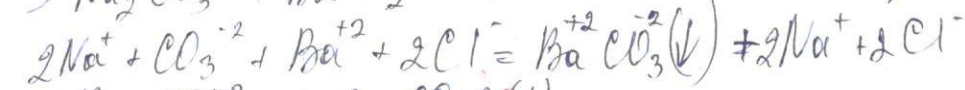
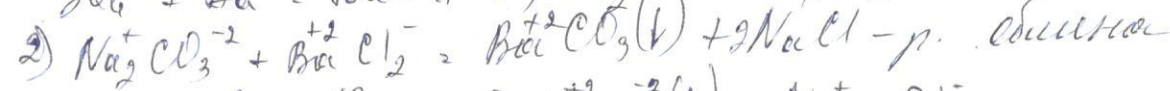
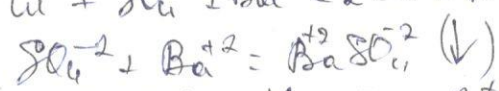
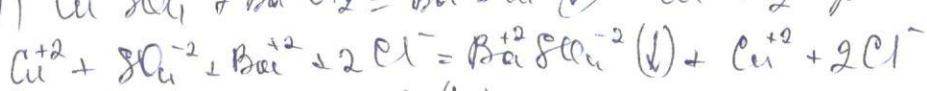
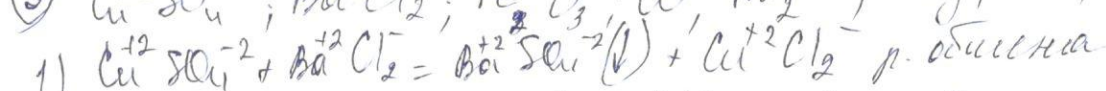
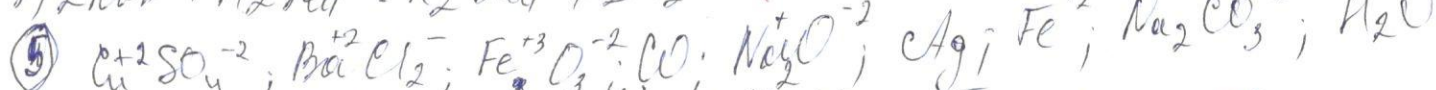
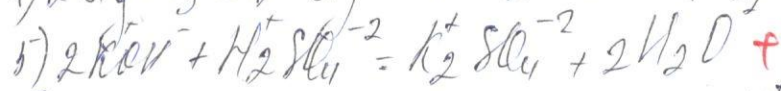
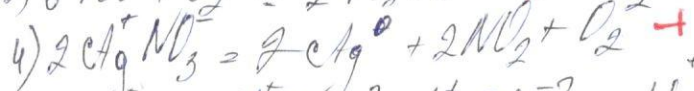
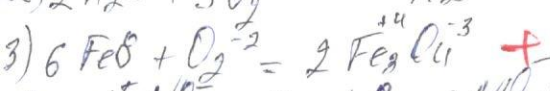
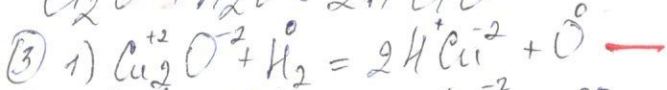


② в низших оксидах валентность всегда = 1

$$W = 18,4\% = 0,184$$

$$\text{Mr}(\text{O}) = \frac{16}{0,184} = 87$$

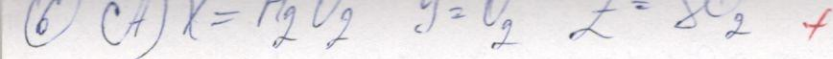
$$\text{Ar}(\text{O}) = (87 - 16) : 2 = 35,5 \quad W(\text{O}) = \frac{16 \cdot 4}{183} \cdot 100\% \approx 34,2\%$$



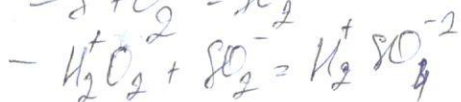
+105.

85.

45



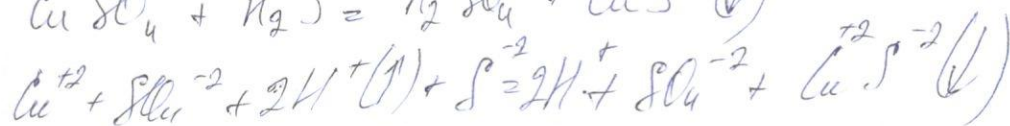
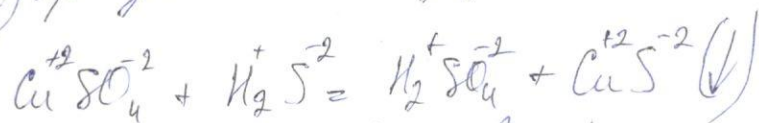
Б) -



В) если в этот раствор добавить кусочек цинка, то цвет раствора изменится  $\Rightarrow$  реакция произошла. +

Г) пероксид водорода - это окислитель водорода, используемый для обработки ртутных +

Д) пропускание водорода через  $Cu^{+2}SO_4^{-2}$



(+105)

$Z = 10 + 8 + 4 + 10 = 325$