

1. Na

 $M(\text{Na}) = 22,9 \text{ г/моль}$  $\text{NaOH}$  - гидроксид натрия (щелочь)  $\text{Na}^+ + \text{OH}^-$  $\text{NaCl}$  - соль  $\text{Na}^+ + \text{Cl}^-$ 

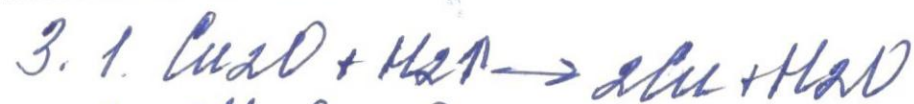
⊖

2. Хлорный оксид:  $\text{Cl}_2\text{O}$ Бромный оксид:  $\text{Cl}_2\text{O}_7$ 

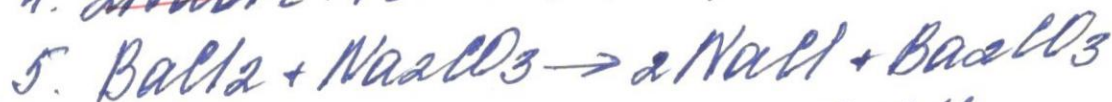
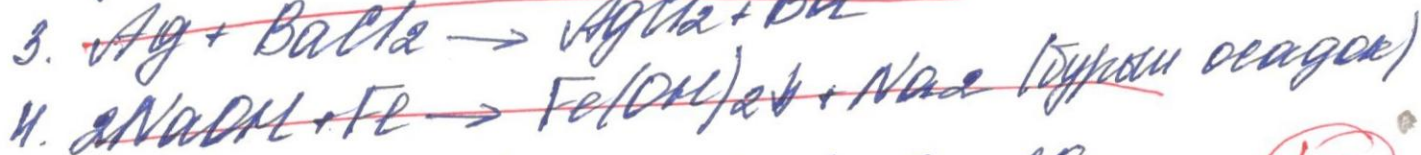
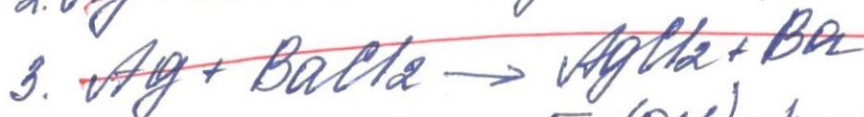
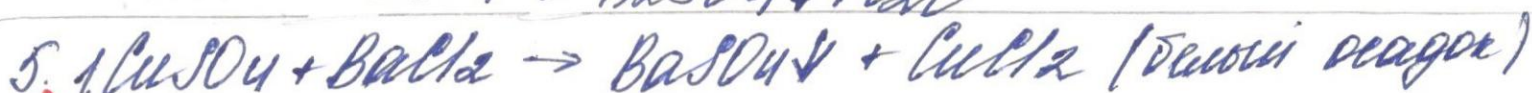
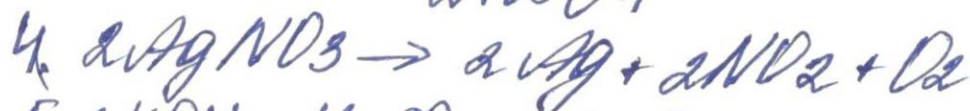
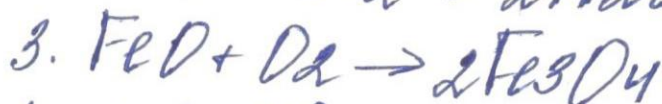
$$w(\text{O}) = 16 \cdot 7 : 183 \cdot 100\% = 61,22\%$$



+105



+105



+15



$$4\text{CH}_4 = 53,6\% \quad +15$$

 $\Sigma 255$